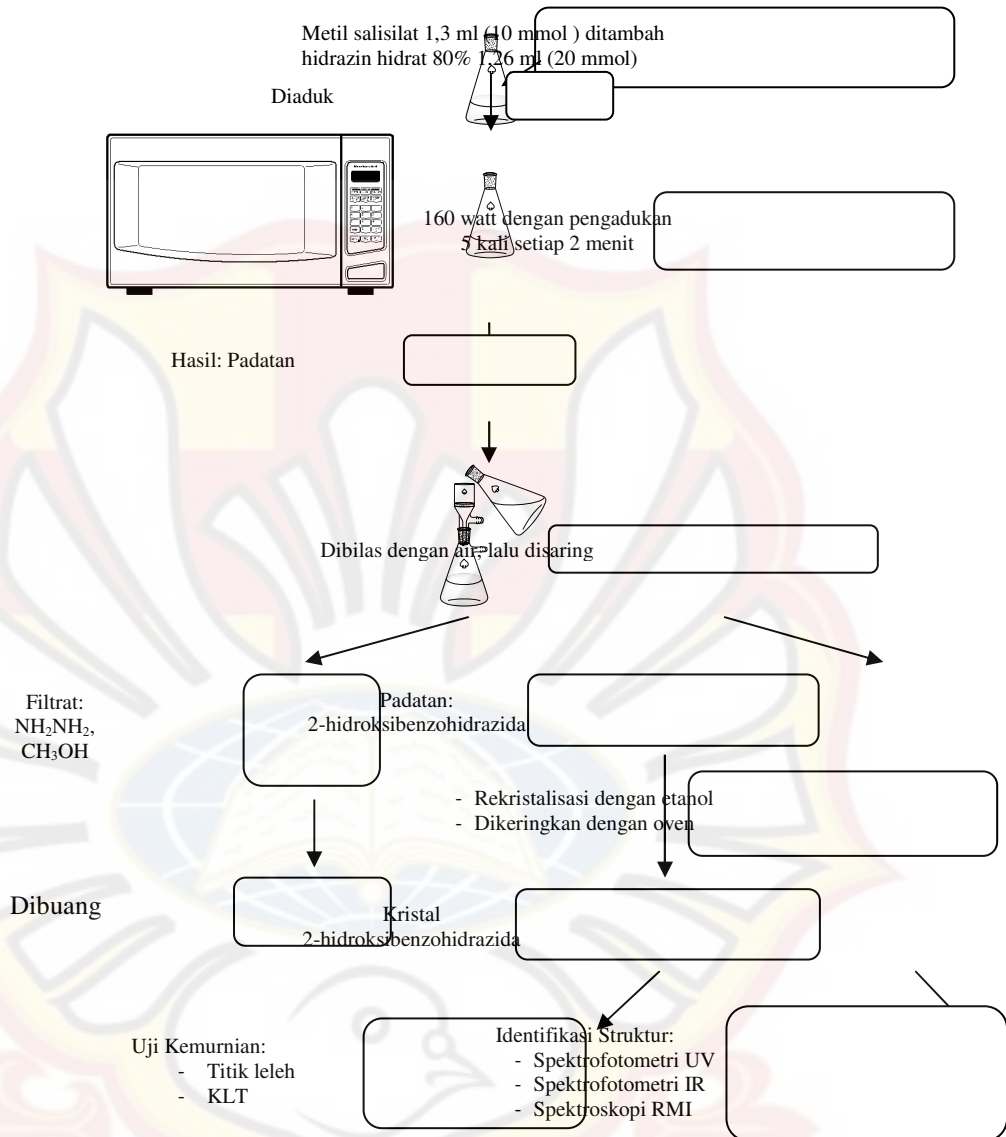
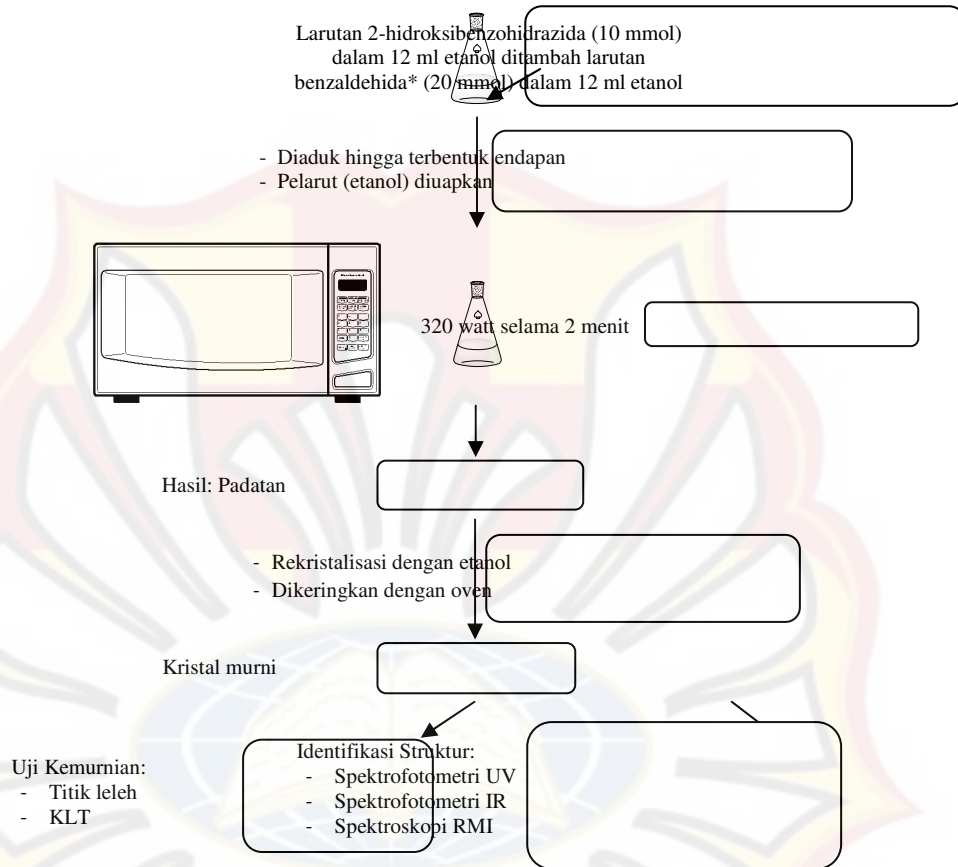


LAMPIRAN A

SKEMA SINTESIS 2-HIDROKSIBENZOHIDRAZIDA



LAMPIRAN B
SKEMA SINTESIS N'-BENZILIDEN-2-HIDROKSIBENZOHI-
DRAZIDA, N'-(2-KLOROBENZILIDEN)-2-HIDROKSIBENZO-
HIDRAZIDA, N'-(2,4-DIKLOROBENZILIDEN)-2-HIDROKSI-
BENZOHI-DRAZIDA



* keterangan:

- Untuk sintesis N'-benziliden-2-hidroksibenzohidrazida menggunakan benzaldehida sebanyak 2,0 ml (20 mmol)
- Untuk sintesis N'-(2-klorobenziliden)-2-hidroksibenzohidrazida menggunakan 2-klorobenzaldehida sebanyak 2,26 ml (20 mmol)
- Untuk sintesis N'-(2,4-diklorobenziliden)-2-hidroksibenzohidrazida menggunakan 2,4-diklorobenzaldehida sebanyak 3,5 g (20 mmol)

LAMPIRAN C
PERHITUNGAN BERAT TEORITIS
2-HIDROKSIBENZOHIDRAZIDA

1. Metil salisilat (BM = 152,15; BJ = 1,184)

volume metil salisilat = 1,3 ml

$$\text{mmol metil salisilat} = \frac{1,3 \times 1,184}{152,15} \times 1000 = 10,12 \text{ mmol}$$

2. Hidrazin hidrat (BM = 50,05; BJ = 1,03)

volume hidrazin hidrat = 1,2 ml

$$\text{mmol hidrazin hidrat} = \frac{1,2 \times 1,03}{50,05} \times 1000 = 24,70 \text{ mmol}$$

Metil salisilat	+	hidrazin hidrat	→	2-hidroksibenzohidrazida
10,12 mmol		24,70 mmol		-
10,12 mmol		10,12 mmol		10,12 mmol
0 mmol		14,58 mmol		10,12 mmol

Jadi didapatkan mmol teoritis 2-hidroksibenzohidrazida = 10,12 mmol

3. 2-hidroksibenzohidrazida (BM = 152,15)

mmol teoritis = 10,12 mmol

$$\text{berat teoritis} = 10,12 \times 152,15 = 1539,76 \text{ mg} = 1,54 \text{ gram}$$

LAMPIRAN D
PERHITUNGAN PERSENTASE HASIL SINTESIS
2-HIDROKSIBENZOHIDRAZIDA

Berat molekul 2-hidroksibenzohidrazid = 152,15

1. Sintesis I

berat praktis = 1,19 gram

$$\text{mmol praktis} = \frac{1,19}{152,15} \times 1000 = 7,82 \text{ mmol}$$

mmol teoritis = 10,12 mmol

$$\text{persentase hasil} = \frac{7,82}{10,12} \times 100\% = 77,27\%$$

2. Sintesis II

berat praktis = 1,21 gram

$$\text{mmol praktis} = \frac{1,21}{152,15} \times 1000 = 7,95 \text{ mmol}$$

mmol teoritis = 10,12 mmol

$$\text{persentase hasil} = \frac{7,95}{10,12} \times 100\% = 78,55\%$$

3. Sintesis III

berat praktis = 1,15 gram

$$\text{mmol praktis} = \frac{1,15}{152,15} \times 1000 = 7,56 \text{ mmol}$$

mmol teoritis = 10,12 mmol

$$\text{persentase hasil} = \frac{7,56}{10,12} \times 100\% = 74,70\%$$

$$\text{Persentase hasil rata-rata} = \frac{77,27\% + 78,55\% + 74,70\%}{3} = 76,84\% \approx 77\%$$

LAMPIRAN E
PERHITUNGAN BERAT TEORITIS
N'-BENZILIDEN-2-HIDROKSIBENZOHIDRAZIDA

1. 2-hidroksibenzohidrazid

mmol teoritis = 10,12 mmol

2. Benaldehida (BM = 106,12, BJ = 1,046)

volume benaldehida = 2,0 ml

mmol benaldehida = $\frac{2,0 \times 1,046}{106,12} \times 1000 = 19,71 \text{ mmol}$

2-hidroksibenzohidrazida + benaldehida		→ N'-benziliden-2-hidroksibenzohidrazida
10,12 mmol	19,71 mmol	-
10,12 mmol	10,12 mmol	10,12 mmol
0 mmol	9,59 mmol	10,12 mmol

Jadi didapatkan mmol teoritis N'-benziliden-2-hidroksibenzohidrazida = 10,12 mmol

3. N'-benziliden-2-hidroksibenzohidrazida (BM = 240,26)

mmol teoritis = 10,12 mmol

berat teoritis = $10,12 \times 240,26 = 2431,43 \text{ mgram} = 2,43 \text{ gram}$

LAMPIRAN F
PERHITUNGAN PERSENTASE HASIL SINTESIS
N'-BENZILIDEN-2-HIDROKSIBENZOHIDRAZIDA

Berat molekul N'-benziliden-2-hidroksibenzohidrazida = 240,26

1. Sintesis I

berat praktis = 1,71 gram

$$\text{mmol praktis} = \frac{1,71}{240,26} \times 1000 = 7,12 \text{ mmol}$$

mmol teoritis = 10,12 mmol

$$\text{persentase hasil} = \frac{7,12}{10,12} \times 100\% = 70,36\%$$

2. Sintesis II

berat praktis = 1,78 gram

$$\text{mmol praktis} = \frac{1,78}{240,26} \times 1000 = 7,41 \text{ mmol}$$

mmol teoritis = 10,12 mmol

$$\text{persentase hasil} = \frac{7,41}{10,12} \times 100\% = 73,22\%$$

3. Sintesis III

berat praktis = 1,73 gram

$$\text{mmol praktis} = \frac{1,73}{240,26} \times 1000 = 7,20 \text{ mmol}$$

mmol teoritis = 10,12 mmol

$$\text{persentase hasil} = \frac{7,20}{10,12} \times 100\% = 71,15\%$$

$$\text{Persentase hasil rata-rata} = \frac{70,36\% + 73,22\% + 71,15\%}{3} = 71,58\% \approx 72\%$$

LAMPIRAN G
UJI STATISTIK PERSENTASE HASIL SINTESIS

1. Uji *independent samples t-test* terhadap persentase hasil sintesis N'-benziliden-2-hidroksibenzohidrazida dan N'-(2-klorobenziliden)-2-hidroksibenzohidrazida.

Independent Samples Test

		Levene's Test for Equality of Variances		t-test for Equality of Means						95% Confidence Interval of the Difference	
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference		Lower	Upper
Randemen	Equal variances assumed	,000	1,000	2,940	4	,042	3,66667	1,24722		,20383	7,12950
	Equal variances not assumed			2,940	4,000	,042	3,66667	1,24722		,20383	7,12950

2. Uji *independent samples t-test* terhadap persentase hasil sintesis N'-benziliden-2-hidroksibenzohidrazida dan N'-(2,4-diklorobenziliden)-2-hidroksibenzohidrazida.

Independent Samples Test

		Levene's Test for Equality of Variances		t-test for Equality of Means						95% Confidence Interval of the Difference	
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference		Lower	Upper
Randemen	Equal variances assumed	,500	,519	6,485	4	,003	9,66667	1,49071		5,52779	13,80555
	Equal variances not assumed			6,485	3,670	,004	9,66667	1,49071		5,37666	13,95667

3. Uji *independent samples t-test* terhadap persentase hasil sintesis N'-(2-klorobenziliden)-2-hidroksibenzohidrazida dan N'-(2,4-diklorobenziliden)-2-hidroksibenzohidrazida.

Independent Samples Test

		Levene's Test for Equality of Variances		t-test for Equality of Means						95% Confidence Interval of the Difference	
		F	Sig.	t	df	Sig. (2-tailed)	Mean Difference	Std. Error Difference		Lower	Upper
VAR00002	Equal variances assumed	,500	,519	4,025	4	,016	6,00000	1,49071		1,86112	10,13688
	Equal variances not assumed			4,025	3,670	,019	6,00000	1,49071		1,70999	10,29001